

Growth and Metal Removal Efficiency of The Green Algae *Schroederia setigera* and *Selenastrum bibrainum* Exposed to Nickel, Zinc, and Cadmium

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INTRODUCTION

Nowadays, more attention has been paid to heavy metal pollution as the major consequence of the rapid development of industrialization and urbanization. Among trace metals, Ni and Zn are essential components while Cd are non-essential for living things [1]. However, when exceeding a certain concentration, these metals could be toxic to organisms. Thus, the occurrence of these metals at high levels could be one of the biggest concerns for the environment, ecosystem, and human health.

Microalgae play an essential role in aquatic ecosystems such as producing oxygen and being the food source for other organisms in higher trophic levels. Therefore, the adverse effects of metals on microalgae could strongly change the structure of aquatic ecosystems. There have been numerous studies on the detrimental impacts of Zn, Ni, and Cd on the growth, photosynthesis and morphological abnormalities of microalgae [2]. On the other hand, due to the rapid growth and bio-absorption capacity, many microalgae have been known as potential organisms for metal removal with high efficiency and a friendly mean to the environment. Various investigations on the growth of green algae (e.g. *Scenedesmus*, *Chlorella*) and their metal removal capacity were reported [3]. However, the responses of many green algae (e.g. *Schroederia*, *Selenastrum*) under the exposure to heavy metals have not been fully understood. Therefore, this study investigated the growth and heavy metal removal efficiency of the two green algae *Schroederia setigera* and *Selenastrum bibrainum* isolated from Vietnam under the exposures to Zn, Ni, and Cd.

MATERIALS AND METHODS

Materials and Methods

The metal solutions (Ni(NO₃)₂, Zn(NO₃)₂, and Cd(NO₃)₂) at 1000 mg/L (ICP/MS standard analysis) were purchased from Merck (Germany) and used as mother solutions for the experiments.

The algal samples were collected from the Nhieu Loc-Thi Nghe Canal in Hochiminh City. The green algae *S. setigera* Lemmermann 1898 and *S. bibrainum* Reinsch 1866 (Fig. 1) were isolated by the pipetting and washing method [4].

The algae were cultured in the Z8 medium [5] under the laboratory conditions (at 27 ± 1 °C, light intensity around 2500 Lux and light: dark cycle of 12h:12h) [6].

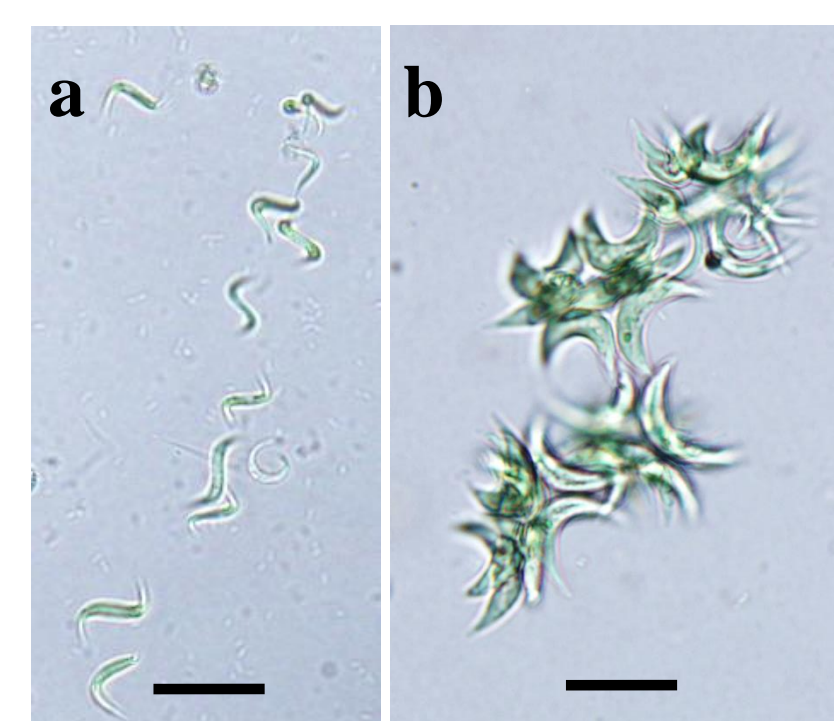


Figure 1 - *Schroederia setigera* (a) and *Selenastrum bibrainum* (b). Scale bars = 20 µm.

Studying the development of the algae upon exposures to three trace metals

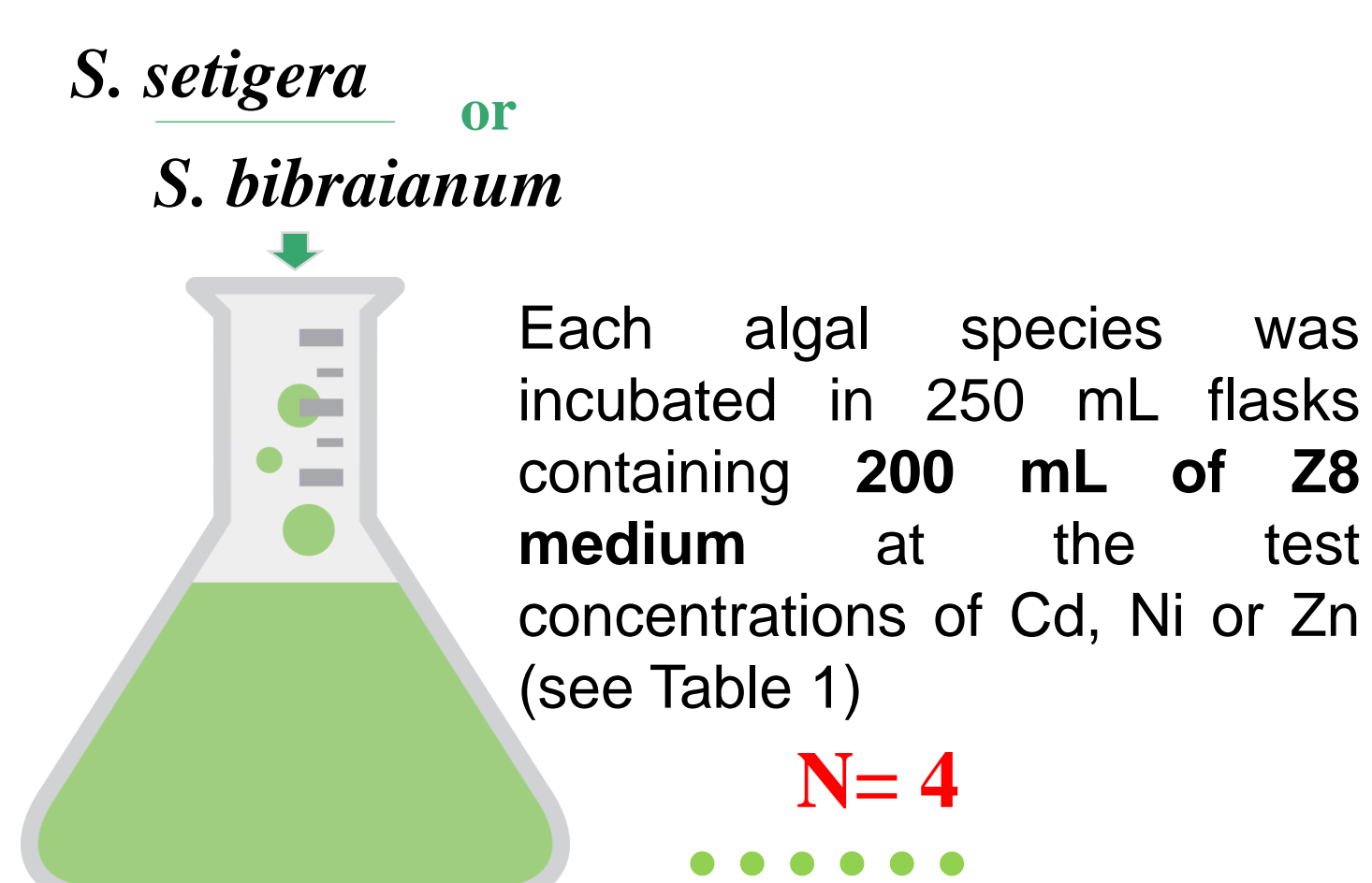


Table 1. Summary of the test concentrations of Cd, Zn, and Ni.

No.	Abbreviations of the exposures	Cd (µg/L)	Zn (µg/L)	Ni (µg/L)
1	Control	0	0	0
2	Cd5	5	-	-
3	Cd200	200	-	-
4	Zn100	-	100	-
5	Zn200	-	200	-
6	Ni100	-	-	100
7	Ni200	-	-	200

There were four replicates with a similar initial density

The tests lasted 18 days.

The algal density in each culturing flask was determined at the start and every two days of the exposures.

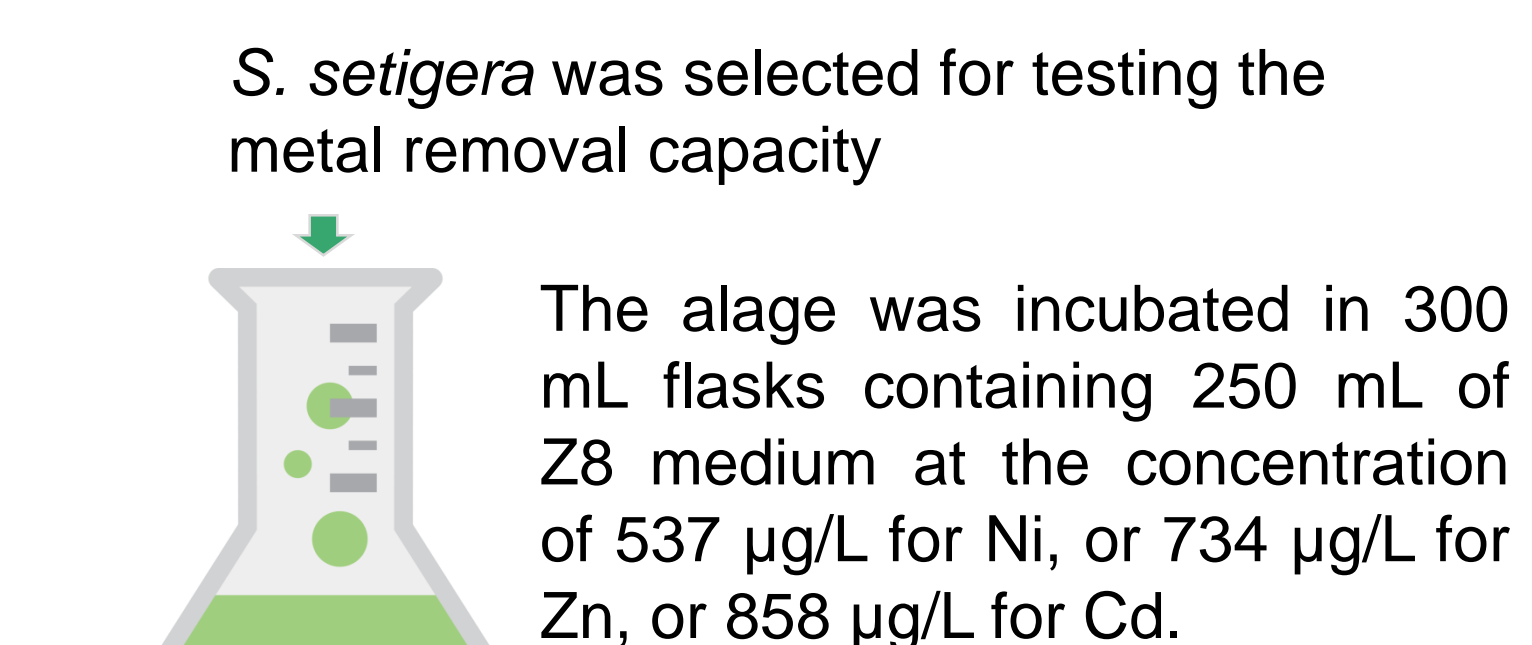
The algae were counted with a Sedgewick Rafter counting chamber (Graticules Optics, England) [6].

Studying the metal removal of the algae upon exposures to three trace metals

This experiment lasted for 16 days.

The metal concentrations were determined at the day 0 (starting day), day 8th, and day 16th (end of the experiment).

The Electrothermal Atomic Absorption Spectrometric Method (PinAAcle 900Z, Perkin Elmer, USA) was used for metal analysis [6].



Three replicates were conducted for each test concentration

Data treatment

The growth rate (R) of microalgae was calculated by the equation of $R = (\ln X_1 - \ln X_2) / (t_1 - t_2)$; where X₁ and X₂ are algal density at time t₁ and t₂ [7].

The formula of $(E\%) = 100 \times (M_1 - M_2) / M_1$ was applied to determine the metal removal efficiency by algae; where M₁ and M₂ are metal concentration at the beginning and the end of the test.

RESULTS AND DISCUSSION

Development of the algae under exposure to trace metals

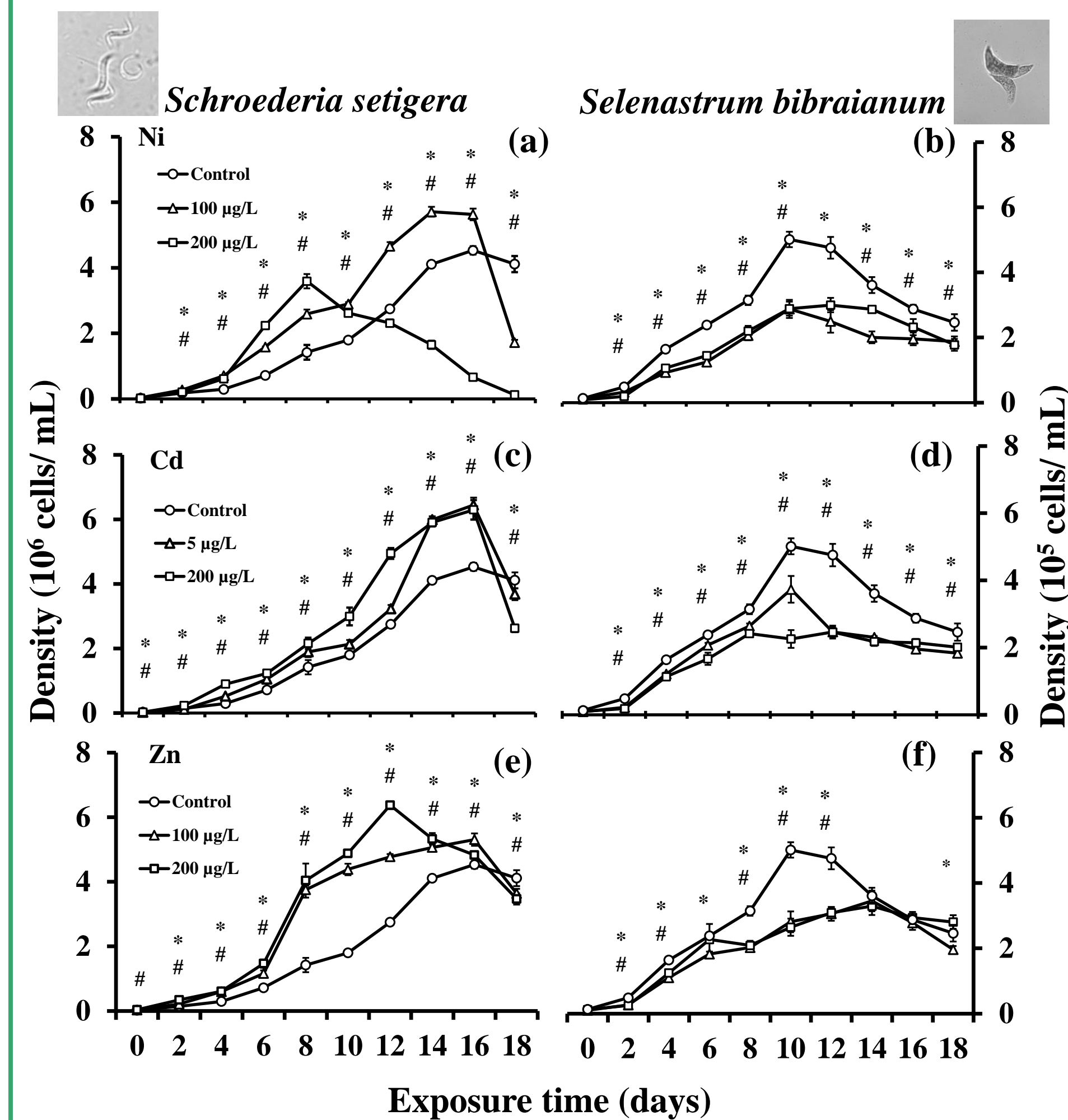


Figure 2 – The density of *Schroederia setigera* and *Selenastrum bibrainum* under exposure to Ni (a, b), Cd (c, d), and Zn (e, f) during 18 days

The results showed that exposures to three trace metals caused inhibitory on the growth of *S. bibrainum*.

Seriously, the cell size of *S. bibrainum* in the 200 µg Zn/L treatment was around two times smaller than the control ($9.2 \pm 2.5 \mu\text{m}$ compared to $18.9 \pm 2.5 \mu\text{m}$; Fig. 3).

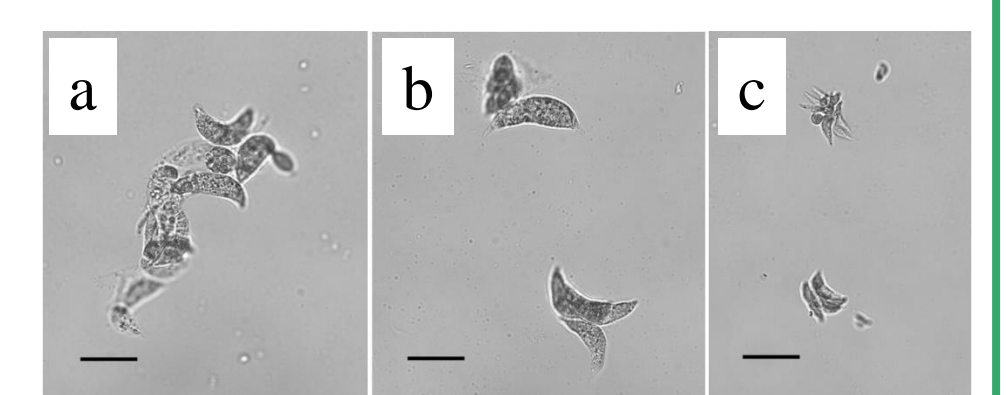


Figure 3 – *Selenastrum bibrainum* in control (a, b) and Zn200 treatment (c). Scale bars = 20 µm.

In contrast, Zn and Cd at the concentration up to 200 µg/L did not inhibit the growth of *S. setigera* over 18 days of exposure.

Similarly, the *S. setigera* also grew well during the first 8 days exposed to Ni at 200 µg/L before inhibited after that.

Previous studies indicated that Ni_(100 µg/L) could impair the chlorophyll content and carbon assimilation, while both Zn and Cd strongly affect the photosynthesis and growth of microalgae [8] that help to explain the inhibitory of on the growth of *S. bibrainum* in our study.

In contrast, Le et al. (2019) found a faster development of the green alga *Scenedesmus protuberans* treated with higher than 100 µg Cd/L [9]. The alga *Tetraselmis* sp. could have a Zn tolerance up to 250 µg/L [10]. Besides, the toxic effects of heavy metals on the algae were dependent on the concentrations, exposure time, and the sensitivity of species. These results could be great support to explain the tolerance of *S. setigera* to the metals (Cd, Ni, Zn) in our study.

Metal removal by the alga *Schroederia setigera*

Table 2 – Metal removal efficiency of *Schroederia setigera*

Metals	Metal concentrations (µg/L)			Uptake ratio (%)	
	Starting day	After 8 days	After 16 days	After 8 days	After 16 days
Ni	537	485	472	10	12
Zn	734	292	252	60	66
Cd	858	750	712	14	18

The Zn removal efficiency in our study is in line with the study of Kutlu & Mutlu (2017), in which the green alga *Dunaliella* sp. had high efficiency in the Zn removal (up to 85%), but could only reduce less than 10% of Cd out of the test solution [11].

Besides, Vo et al. (2020) indicated that the diatom *Cyclotella* sp. isolated from Vietnam had a high capacity for both Cr and Cd removal, reaching 99-100% [3]. Wang & Wood (1984) showed the alga *Scenedesmus* sp. could remove 65% of Ni after 24 days [12].

The difference in Cd and Ni removal efficiency in our study compared to the previous studies may be due to different algae species were used [11]. Moreover, the metal removal capacity of the algae could be affected by the characteristics of the algal culture (e.g. the chemical/ physical characteristic of the medium, the cultural conditions, exposure times, etc.) [12].

CONCLUSIONS

Our study showed the different responses of two green algae *S. setigera* and *S. bibrainum* isolated from Vietnam under exposure to Ni, Cd, and Zn at the concentrations between 5 and 200 µg/L. Although these metals did not cause inhibitory on the growth of *S. setigera*, the growth of *S. bibrainum* was inhibited during 18 days of the exposure.

Our results indicated that Ni and Zn at the concentrations within the permissible limit according to Vietnam Technical Regulation for surface water quality (QCVN 08-MT:2015/BTNMT) could cause negative effects on the growth of the *S. bibrainum*. Therefore, further investigations on the safe concentrations of these metals should be suggested to protect the ecosystem balance.

On the other hand, the alga *S. setigera* had high efficiency in Zn removal, thus using *S. setigera* as an organism for biotechnology to treat Zn in wastewater is recommended.

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Acknowledgement: This research is funded by Vietnam National University Hochiminh City (VNU-HCM) under grant number C2020-20-41.